

Figure 5. Bronsted plot of $\log k_{\mathrm{ROH}}$ or $\log k_{\mathrm{RO}}-\mathrm{vs}$. $\mathrm{p} K_{\mathrm{a}}$ of the conjugate acid. $\mathrm{p} K_{\mathrm{a}}$ data: phenol, Arnett, E. M. Prog. Phys. Org. Chem. 1963, 1, 223; alcohol, Perdoncin, G.; Scorrano, G. J. Am. Chem. Soc. 1977, 99, 6983; phenoxide, this research; alkoxide, ref 1. In the phenol-phenoxide pairs, the dimethyl system (3) is indicated by open squares.

They observed catalysis by borate, but not by acetate, and presented a three-point Brønsted plot ( $\mathrm{H}_{2} \mathrm{O}$, borate, $\mathrm{OH}^{-}$) with $\beta=0.25$. More recently, our studies with the cyclopentanol derivative 18 established general base catalysis by a variety of


18


19
oxyanion buffers with $\beta=0.27$. Subsequent studies by Irie and Tanida ${ }^{17}$ with 19 extended the observations of general base catalysis. An unusual finding in the latter study was the apparent catalysis by general acids (e.g., acetic acid); similar general-acid catalysis has not been observed in our studies of 18. The present work has extended the observation of general-base catalysis of $\mathrm{S}_{\mathrm{N}} 2$ reactions further by demonstrating the susceptibility of phenols such as $\mathbf{1}$ and 3 to this type of catalysis. Previous studies by Borchardt and Cohen ${ }^{3}$ failed to provide evidence for buffer catalysis in cyclization reactions of molecules of type 20. The

$$
\mathrm{X}=-\mathrm{OSO}_{2} \mathrm{CH}_{3},-\mathrm{OPO}(\mathrm{OEt})_{2},-\mathrm{COOCl}_{3}
$$

main difference between 20 and the phenols ( 1 and 3 ) studied in the present work is the nature of the leaving group. The leaving groups in $\mathbf{2 0}$ are all negatively charged (from a neutral starting material), whereas the leaving group in $\mathbf{1}$ and $\mathbf{3}$ is neutral (from a previously charged starting material). Although slightly different reaction conditions prevailed in the studies with $\mathbf{2 0}$ ( $40 \%$ dioxane $-60 \% \mathrm{H}_{2} \mathrm{O}$ ) vs. the present work ( $100 \% \mathrm{H}_{2} \mathrm{O}$ ), it is not clear why no general-base catalysis was observed in the reactions of 20. Additional research will be required to deliniate structural features which lead to general catalysis in $\mathrm{S}_{\mathrm{N}} 2$ reactions. As part of this research effort, we have recently completed the synthesis of compounds which are similar to 3 but contain a poorer alkyl thioether leaving group (e.g., $5^{\prime}$-(methylthio)adenosine).

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Registry No. 1, 81246-41-5; 2, 81246-42-6; 3, 81246-43-7; 4, 254-04-6; 5, 40614-27-5; 6, 1849-36-1; 7, 81246-44-8; 8, 81246-45-9; 9, 1481-92-1; 10, 81246-46-0; 11, 10493-37-5; 12, 81246-47-1; 13, 81246-48-2; 14, 40614-20-8; 14 1-mesylate, 40614-15-1; 14 1-tosylate, 81255-45-0; 15, 81246-49-3; 16, 81246-50-6; 17, 81246-51-7; 3,4-dihydrocoumarin, 119-84-6.

(17) Irie, T.: Tanida, H. J. Org. Chem. 1979, 44, 325-330.
(18) Bell, R. P. "The Proton in Chemistry", 2nd ed.; Cornell University Press: Ithaca, N.Y., 1973; Chapter 10.

Table I. Orthogonal Coordinates for Standard Nucleic Acid Base Residues (in Angstroms)

|  | neutral |  | protonated |  |  | neutral |  | protonated |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $x$ | $y$ | $x$ | $y$ |  | $x$ | $y$ | $x$ | $y$ |
| Cytosine |  |  |  |  |  | Uracil |  |  |  |
| N1 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | N1 | 0.0000 | 0.0000 |  |  |
| C2 | 0.0000 | 1.3990 | 0.0000 | 1.3809 | C2 | 0.0000 | 1.3791 |  |  |
| N3 | 1.1871 | 2.0550 | 1.2581 | 1.9646 | N3 | 1.2462 | 1.9560 |  |  |
| C4 | 2.3303 | 1.3673 | 2.4287 | 1.2882 | C4 | 2.4662 | 1.3042 |  |  |
| C5 | 2.3500 | -0.0585 | 2.3668 | -0.1236 | C5 | 2.3823 | -0.1333 |  |  |
| C6 | 1.1739 | -0.6943 | 1.1614 | -0.7115 | C6 | 1.1787 | -0.7172 |  |  |
| O2 | -1.0799 | 2.0023 | -1.0098 | 2.0494 | O 2 | -1.0194 | 2.0456 |  |  |
| N4 | 3.4753 | 2.0576 | 3.5603 | 1.9542 | O4 | 3.5058 | 1.9558 |  |  |
| Adenine |  |  |  |  | Guanine |  |  |  |  |
| N1 | -2.3548 | 1.2363 | -2.3151 | 1.1999 | N1 | -2.3158 | 1.1867 | -2.3171 | 1.1980 |
| C2 | -2.1991 | -0.0930 | -2.1996 | -0.1571 | C2 | -2.2136 | -0.1842 | -2.2130 | -0.1689 |
| N3 | -1.0733 | -0.8054 | -1.0758 | -0.8223 | N3 | -1.0615 | -0.8426 | $-1.0562$ | -0.8228 |
| C4 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | C4 | 0.0000 | 0.0000 | 0.0000 | 0.0000 |
| C5 | 0.0000 | 1.3817 | 0.0000 | 1.3853 | C5 | 0.0000 | 1.3772 | 0.0000 | 1.3721 |
| C6 | -1.2563 | 2.0196 | -1.2441 | 2.0381 | C6 | -1.2362 | 2.0659 | -1.2278 | 2.0873 |
| N7 | 1.2954 | 1.8718 | 1.2865 | 1.8793 | N7 | 1.2986 | 1.8698 | 1.3213 | 1.7829 |
| C8 | 2.0408 | 0.7921 | 2.0435 | 0.8027 | C8 | 2.0457 | 0.8010 | 2.0877 | 0.6984 |
| N9 | 1.3246 | -0.3724 | 1.3156 | -0.3672 | N9 | 1.3265 | -0.3697 | 1.3124 | -0.3934 |
| N6 | -1.4074 | 3.3480 | -1.4243 | 3.3477 | N2 | -3.3684 | -0.8660 | -3.3726 | -0.8464 |
|  |  |  |  |  | 06 | -1.4337 | 3.2891 | -1.4142 | 3.2980 |

Cambridge Crystallographic Database. ${ }^{6}$ The new mean values agree quite well with those reported by Voet and Rich, but there are occasional discrepancies of up to $0.02-0.03 \AA$ for bond lengths and $1-3^{\circ}$ for valence angles. These discrepancies are statistically significant when compared with the standard errors of the means, which average about $0.004 \AA$ for bond lengths and $0.2^{\circ}$ for valence angles. ${ }^{4}$ We have therefore used the redetermined average dimensions to calculate orthogonal coordinates for standard nucleic acid base residues. The results are reported herein.

## Methods and Results

The coordinates for each residue (protonated and neutral cytosine, adenine, and guanine residues, and the neutral uracil residue; Figure 1) were calculated by minimization of the expression

$$
\sum_{i}\left(l_{i}-\bar{l}_{i}\right)^{2} / \sigma_{i}^{2}+\sum_{j}\left(\theta_{j}-\bar{\theta}_{j}\right)^{2} / \sigma_{j}^{2}
$$

$\sum_{i}$ and $\sum_{j}$ represent summations over all the bond lengths and valence angles, respectively. $l_{i}$ is the value of the $i$ th bond length, and $\bar{l}_{i}$ and $\sigma_{i}$ are, respectively, the mean and standard deviation of the bond length in the crystal structures. Similarly, $\theta_{j}$ is the value of the $j$ th valence angle, and $\bar{\theta}_{j}$ and $\sigma_{j}$ are its mean and standard deviation in the crystal structures. The residues were assumed to be precisely planar.

The coordinates are listed in Table I. ${ }^{7}$ The dimensions of the standard residues are summarized in Table II, together with the mean dimensions observed in the crystal structures. ${ }^{4}$ The maximum discrepancy for bond lengths, $\left|l_{i}-\bar{l}_{i}\right|_{\max }$, is $0.001 \AA$, while the corresponding figure for valence angles is $0.1^{\circ}$. These discrepancies are within the uncertainties of the observed average dimensions.

## Base Planarity

The similarity between the dimensions of the standard residues and the average dimensions observed in the crystal structures suggests that the assumption of exact base planarity is justified. However, at the request of one of the referees we examined the

[^0]

Cytosine.


Adenine.


Guanine.


Uracil.

Figure 1. Atom labeling scheme for nucleic acid base residues.
validity of this assumption in detail. From the original sample of 90 crystal structures, we selected 67 that crystallize in Sohncke groups ${ }^{8}$ and in which the chirality of the nucleoside or nucleotide molecule can be inferred from the known configuration of the sugar residue. Calculations of the furanose residue torsion angles revealed that in five cases ${ }^{9}$ the configuration at the $\mathrm{C}^{\prime}$ atom was that of an L sugar, although the authors had named the sugar as D. ${ }^{15}$ We therefore reversed the signs of all the torsion angles in these molecules. We then computed the means and standard errors
(8) Space groups in which homochiral molecules can crystallize.
(9) Arabinofuranosyladenine hydrochloride ${ }_{r}^{10}$ sodium deoxyadenosine $5^{\prime}$ phosphate hexahydrate, ${ }^{11}$ deoxyadenosine monohydrate, ${ }^{12}$ 5-hydroxymethyl-$2^{\prime}$-deoxyuridine, ${ }^{13}$ and the $1: 1$ salicylic acid-cytidine complex. ${ }^{14}$
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(11) Reddy, B. S.; Viswamitra, M. A. Acta Crystallogr., Sect. B. 1975, B3l, 19-26.
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(13) Birnbaum, G. I.; Deslauriers, R.; Lin, T.-S.; Shiau, G. T.; Prusoff, W. H. J. Am. Chem. Soc. 1980, 102, 4236-4241.
(14) Tamura, C.; Yoshikawa, M.; Sato, S.; Hata, T. Chem. Lett. 1973, 1221-1224.
(15) The procedure used to infer the configuration of the sugar residue was that of Murray-Rust, P.; Motherwell, S. Acta Crystallogr., Sect. B 1978, B34, 2534-2546.

Table II. Dimensions of Standard Nucleic Acid Base Residues (Distances in Angstroms, Angles in Degrees)

|  | neutral |  | protonated |  |  | neutral |  | protonated |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | calcd ${ }^{\alpha}$ | obsd | calcd | obsd |  | calcd | obsd | calcd | obsd |
| Cytosine |  |  |  |  | Uracil |  |  |  |  |
| N1-C2 | 1.399 | 1.399 (4) | 1.381 | 1.381 (2) | N1-C2 | 1.379 | 1.379 (2) |  |  |
| $\mathrm{C} 2-\mathrm{N} 3$ | 1.356 | 1.356 (3) | 1.387 | 1.387 (2) | $\mathrm{C} 2-\mathrm{N} 3$ | 1.373 | 1.373 (2) |  |  |
| N3-C4 | 1.334 | 1.334 (2) | 1.352 | 1.352 (2) | N3-C4 | 1.383 | 1.383 (2) |  |  |
| C4-C5 | 1.426 | 1.426 (4) | 1.413 | 1.413 (3) | C4-C5 | 1.440 | 1.440 (2) |  |  |
| C5-C6 | 1.337 | 1.337 (2) | 1.341 | 1.341 (2) | C5-C6 | 1.338 | 1.338 (2) |  |  |
| C6-N1 | 1.364 | 1.364 (2) | 1.362 | 1.362 (3) | C6-N1 | 1.380 | 1.380 (2) |  |  |
| $\mathrm{C} 2-\mathrm{O} 2$ | 1.237 | 1.237 (2) | 1.211 | 1.211 (2) | C2-O2 | 1.218 | 1.218 (2) |  |  |
| C4-N4 | 1.337 | 1.337 (4) | 1.313 | 1.313 (3) | C4-O4 | 1.227 | 1.227 (2) |  |  |
| C6-N1-C2 | 120.6 | 120.6 (1) | 121.5 | 121.5 (1) | C6-N1-C2 | 121.3 | 121.3 (1) |  |  |
| $\mathrm{N} 1-\mathrm{C} 2-\mathrm{N} 3$ | 118.9 | 118.9 (2) | 114.9 | 114.9 (2) | N1-C2-N3 | 114.8 | 114.8 (1) |  |  |
| C2-N3-C4 | 120.0 | 120.0 (2) | 125.1 | 125.1 (2) | $\mathrm{C} 2-\mathrm{N} 3-\mathrm{C} 4$ | 127.0 | 127.0 (1) |  |  |
| N3-C4-C5 | 121.8 | 121.8 (2) | 117.5 | 117.5 (2) | N3-C4-C5 | 114.8 | 114.7 (2) |  |  |
| $\mathrm{C} 4-\mathrm{C} 5-\mathrm{C} 6$ | 117.6 | 117.6 (2) | 118.5 | 118.5 (1) | C4-C5-C6 | 119.2 | 119.2 (2) |  |  |
| $\mathrm{C} 5-\mathrm{C} 6-\mathrm{N} 1$ | 121.0 | 121.0 (2) | 122.5 | 122.5 (1) | C5-C6-N1 | 122.8 | 122.8 (2) |  |  |
| N1-C2-O2 | 119.2 | 119.2 (2) | 123.5 | 123.5 (2) | $\mathrm{N} 1-\mathrm{C} 2-\mathrm{O} 2$ | 123.2 | 123.2 (2) |  |  |
| $\mathrm{N} 3-\mathrm{C} 2-\mathrm{O} 2$ | 121.9 | 121.9 (2) | 121.6 | 121.6 (1) | $\mathrm{N} 3-\mathrm{C} 2-\mathrm{O} 2$ | 122.0 | 122.0 (1) |  |  |
| N3-C4-N4 | 117.9 | 117.9 (3) | 119.5 | 119.5 (2) | N3-C4-O4 | 119.8 | 119.8 (1) |  |  |
| $\mathrm{C} 5-\mathrm{C} 4-\mathrm{N} 4$ | 120.3 | 120.3 (2) | 123.0 | 123.0 (3) | C5-C4-O4 | 125.4 | 125.4 (2) |  |  |
| Adenine |  |  |  |  | Guanine |  |  |  |  |
| N1-C2 | 1.338 | 1.338 (3) | 1.362 | 1.362 (4) | N1-C2 | 1.375 | 1.375 (3) | 1.371 | 1.371 (4) |
| C2-N3 | 1.332 | 1.332 (3) | 1.306 | 1.306 (2) | C2-N3 | 1.327 | 1.327 (2) | 1.329 | 1.329 (5) |
| N3-C4 | 1.342 | 1.342 (2) | 1.354 | 1.354 (3) | N3-C4 | 1.355 | 1.355 (2) | 1.339 | 1.339 (7) |
| C4-C5 | 1.382 | 1.382 (2) | 1.385 | 1.385 (6) | C4-C5 | 1.377 | 1.377 (2) | 1.372 | 1.372 (4) |
| C5-C6 | 1.409 | 1.409 (1) | 1.405 | 1.405 (4) | C5-C6 | 1.415 | 1.415 (5) | 1.421 | 1.420 (11) |
| C6-N1 | 1.349 | 1.349 (2) | 1.360 | 1.360 (2) | C6-N1 | 1.392 | 1.393 (2) | 1.406 | 1.406 (12) |
| C5-N7 | 1.385 | 1.385 (2) | 1.378 | 1.378 (2) | C5-N7 | 1.389 | 1.389 (3) | 1.384 | 1.384 (20) |
| N7-C8 | 1.312 | 1.312 (2) | 1.316 | 1.316 (2) | N7-C8 | 1.304 | 1.304 (3) | 1.328 | 1.328 (14) |
| C8-N9 | 1.367 | 1.367 (4) | 1.378 | 1.378 (4) | C8-N9 | 1.374 | 1.374 (4) | 1.339 | 1.339 (7) |
| N9-C4 | 1.376 | 1.376 (2) | 1.366 | 1.366 (3) | N9-C4 | 1.377 | 1.377 (2) | 1.370 | 1.370 (10) |
| C6-N6 | 1.337 | 1.337 (3) | 1.322 | 1.322 (3) | $\mathrm{C} 2-\mathrm{N} 2$ | 1.341 | 1.341 (3) | 1.343 | $1.343(8)$ |
|  |  |  |  |  | C6-06 | 1.239 | 1.239 (5) | 1.225 | $1.225(3)$ |
| $\mathrm{C} 6-\mathrm{N} 1-\mathrm{C} 2$ | 118.8 | 118.8 (2) | 123.2 | 123.2 (2) | C6-N1-C2 | 124.9 | 124.9 (2) | 124.9 | 124.8 (6) |
| $\mathrm{N} 1-\mathrm{C} 2-\mathrm{N} 3$ | 129.0 | 129.0 (1) | 125.5 | 125.5 (2) | N1-C2-N3 | 124.0 | 124.0 (2) | 123.8 | 123.8 (2) |
| C2-N3-C4 | 110.8 | 110.8 (1) | 112.0 | 112.0 (2) | $\mathrm{C} 2-\mathrm{N} 3-\mathrm{C} 4$ | 111.8 | 111.8 (1) | 112.6 | 112.6 (2) |
| N3-C4-C5 | 126.9 | 126.9 (2) | 127.4 | 127.4 (2) | N3-C4-C5 | 128.4 | 128.4 (2) | 127.9 | 127.9 (1) |
| C4-C5-C6 | 116.9 | 116.9 (1) | 117.7 | 117.7 (2) | C4-C5-C6 | 119.1 | 119.1 (1) | 120.2 | 120.2 (3) |
| C5-C6-N1 | 117.6 | 117.6 (1) | 114.3 | 114.3 (3) | C5-C6-N1 | 111.7 | 111.7 (2) | 110.6 | 110.5 (1) |
| $\mathrm{C} 4-\mathrm{C} 5-\mathrm{N} 7$ | 110.7 | 110.7 (1) | 111.0 | 111.0 (2) | $\mathrm{C} 4-\mathrm{C5}-\mathrm{N} 7$ | 110.8 | 110.8 (2) | 107.3 | 107.3 (9) |
| C5-N7-C8 | 103.9 | 103.9 (2) | 104.1 | 104.1 (2) | C5-N7-C8 | 104.2 | 104.2 (3) | 108.0 | 108.0 (2) |
| N7-C8-N9 | 113.8 | 113.8 (2) | 113.0 | 113.0 (2) | N7-C8-N9 | 113.5 | 113.5 (4) | 109.4 | 109.4 (2) |
| C8-N9-C4 | 105.9 | 105.9 (1) | 106.3 | 106.3 (2) | C8-N9-C4 | 106.0 | 106.0 (2) | 108.7 | 108.7 (4) |
| N9-C4-C5 | 105.7 | 105.7 (1) | 105.6 | 105.6 (2) | N9-C4-C5 | 105.6 | 105.6 (1) | 106.7 | 106.7 (7) |
| N3-C4-N9 | 127.4 | 127.4 (1) | 127.0 | 127.0 (4) | N3-C4-N9 | 126.0 | 126.0 (2) | 125.4 | 125.4 (7) |
| C6-C5-N7 | 132.4 | 132.3 (2) | 131.3 | 131.3 (2) | $\mathrm{C} 6-\mathrm{C} 5-\mathrm{N} 7$ | 130.1 | 130.1 (2) | 132.5 | 132.5 (6) |
| N1-C6-N6 | 119.0 | 119.0 (2) | 120.2 | 120.2 (2) | $\mathrm{N} 1-\mathrm{C} 2-\mathrm{N} 2$ | 116.3 | 116.3 (2) | 115.9 | 115.9 (4) |
| C5-C6-N6 | 123.4 | 123.4 (2) | 125.5 | 125.5 (3) | N3-C2-N2 | 119.7 | 119.7 (2) | 120.2 | 120.2 (1) |
|  |  |  |  |  | N1-C6-06 | $120.0$ | 120.0 (2) | 120.5 | 120.5 (13) |
|  |  |  |  |  | C5-C6-06 | 128.3 | 128.3 (2) | 129.0 | 129.0 (12) |

${ }^{a}$ Calcd = value of dimension in standard residuc; obsd = mean value of dimension in crystal structures ${ }^{4}$ (number in parentheses is standard error of mean, in the units of the least significant digit of the preceding number).
of torsion angles involving the base residue atoms. The results are given in Table III. Of the 86 mean torsion angles calculated, all but 5 are within $2 \sigma$ of the values expected for precisely planar base residues, and all but 1 are within $3 \sigma$. This evidence therefore indicates that the mean geometry of each of the base residues is exactly, or almost exactly planar. In order to confirm this, we conducted Hotelling $T^{2}$ significance tests ${ }^{16}$ on four of the residues (protonated and neutral cytosine, neutral adenine, and neutral uracil; insufficient data were available for $T^{2}$ tests on the other three residues). In this procedure, the null hypothesis

$$
\mathrm{H}_{0}: \quad \bar{\tau}=\mu_{0}
$$

was tested against the alternative hypothesis

$$
\mathrm{H}_{1}: \mp \neq \mu_{0}
$$

where $\bar{\tau}$ is the vector of observed mean torsion angles and $\mu_{0}$ is
(16) Morrison, D. F. "Multivariate Statistical Methods", 2nd ed.; McGraw-Hill Kogakusha: Tokyo, 1976; p 128.
the vector of expected mean torsion angles assuming that the average geometry of the residue is precisely planar. The $T^{2}$ values obtained, together with their associated $F$ statistics (degrees of freedom in parentheses), are as follows: protonated cytosine $T^{2}$ $=98.2, F(10,1)=0.982$; neutral cytosine $T^{2}=113.7, F(10,2)$ $=2.068$; neutral adenine $T^{2}=192.9, F(13,1)=1.141$; neutral uracil $T^{2}=9.1, F(10,16)=0.579$. Since the $F$ values are not statistically significant at, or above the $90 \%$ level, we accept the null hypothesis and conclude that the average geometry of each of the residues is precisely planar. ${ }^{17}$

Also given in Table III are the means and standard errors of the absolute values of the torsion angles (i.e., of $|\tau|$ rather than

[^1]Table III. Mean Torsion Angles of Nucleic Acid Base Residues (in Degrees)

|  | neutral |  |  | protonated |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\bar{\tau}(\exp )^{a}$ | $\bar{\tau}$ (obsd) | I/ $\bar{\tau}$ (obsd) | $\bar{\tau}(\mathrm{exp})$ | $\bar{\tau}$ (obsd) | $\sqrt{\text { F }}$ ( (obsd) |
| Cy tosine |  |  |  |  |  |  |
| N1-C2-N3-C4 | 0 | 1.4 (0.8) | 2.3 (0.4) | 0 | -0.6 (1.1) | 2.1 (0.5) |
| C2-N3-C4-C5 | 0 | -1.8(0.7) | 2.5 (0.3) | 0 | -0.7 (0.9) | 1.8 (0.4) |
| N3-C4-C5-C6 | 0 | 1.3 (0.8) | 2.2 (0.5) | 0 | 1.1 (0.9) | 2.2 (0.3) |
| C4-C5-C6-N1 | 0 | -0.5 (0.5) | 1.5 (0.3) | 0 | -0.2 (0.4) | 1.0 (0.2) |
| C5-C6-N1-C2 | 0 | 0.1 (1.0) | 2.4 (0.6) | 0 | -1.1 (0.8) | 1.8 (0.4) |
| $\mathrm{C} 6-\mathrm{N} 1-\mathrm{C} 2-\mathrm{N} 3$ | 0 | -0.6(1.2) | 3.1 (0.6) | 0 | 1.4 (1.3) | 2.5 (0.6) |
| $\mathrm{C} 6-\mathrm{N} 1-\mathrm{C} 2-\mathrm{O} 2$ | 180 | -179.8(1.1) | 177.4 (0.6) | 180 | -178.9 (1.0) | 178.0 (0.5) |
| $\mathrm{C} 4-\mathrm{N} 3-\mathrm{C} 2-\mathrm{O} 2$ | 180 | -179.4 (0.7) | 178.0 (0.3) | 180 | 179.8 (0.9) | 178.2 (0.4) |
| $\mathrm{C} 2-\mathrm{N} 3-\mathrm{C} 4-\mathrm{N} 4$ | 180 | 178.5 (0.8) | 177.5 (0.3) | 180 | 179.4 (0.9) | 178.0 (0.4) |
| $\mathrm{C} 6-\mathrm{C} 5-\mathrm{C} 4-\mathrm{N} 4$ | 180 | -179.0 (0.8) | 177.8 (0.5) | 180 | -179.0 (1.1) | 177.7 (0.4) |
| Uracil |  |  |  |  |  |  |
| N1-C2-N3-C4 | 0 | 0.2 (0.6) | 2.3 (0.3) |  |  |  |
| $\mathrm{C} 2-\mathrm{N} 3-\mathrm{C} 4-\mathrm{C} 5$ | 0 | -0.8 (0.7) | 3.1 (0.3) |  |  |  |
| N3-C4-C5-C6 | 0 | 0.7 (0.7) | 2.8 (0.4) |  |  |  |
| $\mathrm{C} 4-\mathrm{C} 5-\mathrm{C} 6-\mathrm{N} 1$ | 0 | -0.1 (0.4) | 1.4 (0.2) |  |  |  |
| $\mathrm{C} 5-\mathrm{C} 6-\mathrm{N} 1-\mathrm{C} 2$ | 0 | -0.5 (0.5) | 2.3 (0.3) |  |  |  |
| C6-N1-C2-N3 | 0 | 0.4 (0.7) | 2.8 (0.4) |  |  |  |
| C6-N1-C2-O2 | 180 | -179.8 (0.7) | 177.5 (0.4) |  |  |  |
| $\mathrm{C} 4-\mathrm{N} 3-\mathrm{C} 2-\mathrm{O} 2$ | 180 | -179.6 (0.6) | 177.7 (0.3) |  |  |  |
| $\mathrm{C} 2-\mathrm{N} 3-\mathrm{C} 4-\mathrm{O} 4$ | 180 | 179.2 (0.6) | 177.2 (0.3) |  |  |  |
| C6-C5-C4-O4 | 180 | -179.3 (0.7) | 177.4 (0.4) |  |  |  |
| Adenine |  |  |  |  |  |  |
| N1-C2-N3-C4 | 0 | -0.6 (0.4) | 1.1 (0.2) | 0 | -0.2 (0.3) | 0.6 (0.1) |
| $\mathrm{C} 2-\mathrm{N} 3-\mathrm{C} 4-\mathrm{C} 5$ | 0 | 0.1 (0.4) | 1.0 (0.2) | 0 | 1.4 (0.5) | 1.9 (0.3) |
| N3-C4-C5-C6 | 0 | 0.3 (0.4) | 1.1 (0.2) | 0 | -1.4 (0.7) | 2.2 (0.4) |
| C4-C5-C6-N1 | 0 | -0.2 (0.5) | 1.3 (0.2) | 0 | 0.0 (0.7) | 1.6 (0.4) |
| C5-C6-N1-C2 | 0 | -0.3 (0.5) | 1.3 (0.3) | 0 | 1.1 (0.8) | 2.2 (0.4) |
| C6-N1-C2-N3 | 0 | 0.8 (0.5) | 0.9 (0.3) | 0 | -1.1(0.7) | 2.0 (0.4) |
| N9-C4-C5-N7 | 0 | 0.2 (0.3) | 0.8 (0.1) | 0 | 0.0 (0.3) | 0.6 (0.2) |
| C4-C5-N7-C8 | 0 | -0.2 (0.2) | 0.5 (0.1) | 0 | 0.0 (0.3) | 0.6 (0.2) |
| $\mathrm{C} 5-\mathrm{N} 7-\mathrm{C} 8-\mathrm{N} 9$ | 0 | 0.2 (0.3) | 0.8 (0.1) | 0 | 0.0 (0.4) | 0.9 (0.2) |
| N7-C8-N9-C4 | 0 | -0.1 (0.4) | 1.0 (0.2) | 0 | 0.0 (0.3) | 0.8 (0.2) |
| C8-N9-C4-C5 | 0 | -0.1(0.4) | 0.9 (0.2) | 0 | 0.0 (0.3) | 0.6 (0.2) |
| C2-N1-C6-N6 | 180 | 180.0 (0.5) | 178.5 (0.2) | 180 | -179.2 (0.6) | 178.2 (0.3) |
| N7-C5-C6-N6 | 0 | -0.5 (0.5) | 1.6 (0.2) | 0 | -0.2 (0.5) | 1.1 (0.3) |
| Guanine |  |  |  |  |  |  |
| $\mathrm{N} 1-\mathrm{C} 2-\mathrm{N} 3-\mathrm{C} 4$ | 0 | -0.2 (0.8) | 1.7 (0.4) | 0 | -0.4 (1.7) | 2.0 (1.0) |
| C2-N3-C4-C5 | 0 | -1.2(1.0) | 1.6 (0.5) | 0 | 0.9 (2.1) | 2.6 (1.2) |
| N3-C4-C5-C6 | 0 | 1.8 (1.0) | 1.9 (0.6) | 0 | -0.7 (0.3) | 0.7 (0.3) |
| C4-C5-C6-N1 | 0 | -0.8(0.5) | 1.3 (0.5) | 0 | 0.0 (2.4) | 3.1 (1.1) |
| C5-C6-N1-C2 | 0 | -0.5 (0.7) | 1.3 (0.1) | 0 | 0.4 (2.8) | 3.3 (1.6) |
| $\mathrm{C} 6-\mathrm{N} 1-\mathrm{C} 2-\mathrm{N} 3$ | 0 | 1.1 (1.0) | 1.8 (0.3) | 0 | -0.2 (0.8) | 1.1 (0.3) |
| $\mathrm{N} 9-\mathrm{C} 4-\mathrm{C} 5-\mathrm{N} 7$ | 0 | 0.1 (0.4) | 0.7 (0.1) | 0 | 0.0 (0.9) | 1.1 (0.5) |
| C4-C5-N7-C8 | 0 | 0.3 (0.3) | 0.6 (0.2) | 0 | -0.4 (0.9) | 1.0 (0.6) |
| C5-N7-C8-N9 | 0 | -0.6 (0.3) | 0.6 (0.2) | 0 | 0.6 (0.6) | 0.6 (0.6) |
| N7-C8-N9-C4 | 0 | 0.6 (0.2) | 0.6 (0.1) | 0 | -0.6 (0.3) | 0.6 (0.3) |
| C8-N9-C4-C5 | 0 | -0.4 (0.3) | 0.7 (0.1) | 0 | 0.4 (0.6) | 0.7 (0.5) |
| C6-N1-C2-N2 | 180 | -178.9 (1.2) | 178.2 (0.3) | 180 | 178.6 (1.0) | 178.6 (1.0) |
| $\mathrm{C} 4-\mathrm{N} 3-\mathrm{C} 2-\mathrm{N} 2$ | 180 | 179.7 (0.9) | 178.3 (0.4) | 180 | -179.2 (1.5) | 178.1 (0.8) |
| $\mathrm{C} 2-\mathrm{N} 1-\mathrm{C} 6-\mathrm{O} 6$ | 180 | 180.0 (0.8) | 178.8 (0.2) | 180 | 179.9 (2.4) | 177.0 (1.0) |
| N7-C5-C6-O6 | 0 | -0.1(0.8) | 0.7 (0.3) | 0 | 1.0 (0.9) | 1.4 (0.5) |

$a^{\bar{\tau}}(\exp )=$ expected mean value of torsion angle, assuming average geometry of residue is planar; $\bar{\tau}$ (obsd) = observed mean value of torsion angle (number in parentheses is standard error of mean); $\sqrt{\boldsymbol{\tau}} \mid$ (obsd) $=$ observed mean of absolute value of torsion angle (number in parentheses is standard error of mean).
$\tau$ ). These figures are based on all 90 of the crystal structures used in the original survey. ${ }^{4}$ The results show that, in any given crystal structure, the torsion angles of a pyrimidine ring are typically within $1-3^{\circ}$ of the values expected for a precisely planar residue. The corresponding range for the torsion angles of the imidazole ring in a purine residue is about $0.5-1.0^{\circ}$. These figures, together with those in Table II, indicate that the geometry of any given nucleic acid base residue is likely to be very similar to that of the corresponding standard residue reported in this paper.

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Registry No. Cytosine, 71-30-7; cytosine, protonated, 20791-98-4; adenine, 73-24-5; adenine, protonated, 18444-01-4; guanine, 73-40-5; guanine, protonated, 51460-61-8; uracil, 66-22-8.

Supplementary Material Available: Table IV, listing references for the 67 crystal structures used in the examination of base planarity ( 5 pages). Ordering information is given on any current masthead page.


[^0]:    (6) Allen, F. H.; Bellard, S.; Brice, M. D.; Cartwright, B. A.; Doubleday, A.; Higgs, H.; Hummelink, T.; Hummelink-Peters, B. G.; Kennard, O.; Motherwell, W. D. S.; Rodgers, J. R.; Watson, D. G. Acta Crystallogr., Sect. B 1979, B35, 2331-2339.
    (7) Coordinates for the $\mathrm{C} 1^{\prime}$ atom of the furanose ring were not derived. We intend to examine the geometry of the glycosidic linkage in a subsequent study.

[^1]:    (17) This result is consistent with either of the following hypotheses: (a) the equilibrium geometry of the isolated base residue is precisely planar; (b) the equilibrium geometry of the isolated base residue is not precisely planar, but deviations from planarity occur with equal facility on either side of the base mean plane. We await the publication of additional structural data before attempting to decide between these two alternatives, but note that in either case the optimum geometry of the standard base residue will be planar.

